

Oasis Academy Lister Park Science Curriculum: Long Term Plan



Year 9

Secure Substantive Knowledge:

- Students build on their Chemistry knowledge of elements and compounds, looking at compounds and formulae used to represent these substances. They also begin to look at how our concept of an atom has changed over time. They look at patterns and how different groups in the periodic table react and bond together and how this can be modelled using different types of diagram. This unit also builds on the knowledge of common reactions in Year 8 so that students are able to predict which substances will be produced in different reactions and how they would prove that these substances have been made. Students are introduced to electrolysis and how this can be used to separate more reactive elements from their ore and create substances like hydrogen and oxygen.
- Within Physics, students take a deeper look at waves and energy transfers, in particular looking at efficiency of these transfers and the GPE, kinetic energy and elastic potential energy store and how calculations allow us to predict the amount of energy that should be held in that store (should a closed system with no energy loss be used!). Students also start to observe and measure physical properties of waves, representing these using diagrams. Students will be introduced to the different types of quantity within science (scalar and vector). They will look at the quantitative effect of different forces on an objects motion and shape and begin to complete more complex calculations and graphical representations of data.
- Building on the use of the microscope in Year 7, students will look in more details at the types of cells. They will begin to discuss how humans use replication of cells to their advantage and how our concept of the human genome has had an impact on our knowledge of inheritance. They will also look more closely at specific types of communicable disease and how new drugs are developed. They will begin to analyse more complex data sets, using this to draw conclusions. Finally, students will go on to look at the brain and eye and how these complex organs in our body function and are susceptible to damage and how our knowledge of science has once again, allowed us to intervene and in lots of cases, identify the issue and put in place solutions.

Secure Disciplinary Knowledge:

- Within this unit, students are given plenty of opportunities to practice representing elements, compounds and general reactions using symbols. They begin to evaluate the limitations of using particular types of model to represent substances. They write their own scientific hypotheses and test these using the evidence to support their conclusions. They begin to identify anomalies and describe how to deal with them. They start to look at more complete relationships on a graph and use lines of best fit to extract data. They develop their bank of scientific diagrams to include wave diagrams and free body diagrams. They build on their use of the microscope in year 7 to discuss the use of one type of microscope over another.
- They continue to complete calculations of increasing difficulty, calculating means, rounding to a given number of decimal places and significant figures and converting a wider range of units without being prompted. There are opportunities to revisit the concept of an evolving scientific knowledge base with discussions around the structure of the atom, developments in microscopes and how these have supported our understanding of scientific concepts. Students also begin to apply their knowledge of science to explain how we have used this to extract resources from the Earth and how this has at times, been wasteful.

Year Group	Autumn 1	Autumn 2	Spring 1	Spring 2	Summer 1	Summer 2
	Topic: Chemistry Fundamentals	Topic: Investigative Chemistry	Topic: Physics - Energy and Waves	Topic: Forces	Topic: Cell Biology	Topic: Communicable Diseases
	Knowledge:	Knowledge:	Knowledge:	Knowledge:	Knowledge:	Knowledge:
	1. Changing states of matter	1. Ionic bonding part 1	1. Energy stores and energy transfers	1. Scalar and vector quantities	1. Types of cells	1. Viral diseases
	2. Atoms and elements	2. Ionic bonding part 2	2. Open and closed systems	2. Types of forces	2. Specialised cells	2. Bacterial diseases
	3. Compounds and formulae	3. Properties of ionic bonding	3. Work done	3. Weight	3. Tissues, organs and systems	3. Fungal and protists
	4. Pure substances and solutions	4. Covalent bonding	4. Power	4. Resultant forces	4. Introducing microscopes	4. Our barriers to diseases
	5. Separation techniques (Demonstration)	5. Properties of covalent structures	5. Efficiency calculations	5. Vector diagrams	5. RP: Using Microscopes	5. The immune system
	6. Chromatography (R.Practical)	6. Giant covalent structures	6. Insulation	6. Speed and velocity	6. Types of microscope	6. Vaccinations
	7. Changing Atomic Theories	7. Metallic Bonding	7. Gravitational potential energy	7. Distance time graphs	7. The Human Genome	7. Medicines
	8. Protons, Neutrons and Electrons	8. Comparing and contrasting types of bonding	8. Kinetic energy	8. Acceleration and deceleration	8. Mitosis and the cell cycle	8. Multiplying bacteria (Separate only)
	9. Electron configuration	9. Word and symbol equations	9. Elastic potential energy	9. Velocity time graphs	9. Incredible stem cells	9. Culturing microorganisms
	10. Isotopes and relative atomic mass	10. Balancing equations	10. Multi-step calculations (GPE/KE/EPE/Efficiency)	10. Terminal Velocity	10. Therapeutic cloning	10. Analysing Antibiotics
	11. The periodic table	11. Conservation of mass	11. Non-renewable resources	11. Newton's first law	11. Asexual reproduction	11. Antibiotic resistance
	12. The modern periodic table	12. Metals and oxygen (Demonstration)	12. Renewable resources	12. Newton's second law	12. Sexual Reproduction and Meiosis	12. Developing new drugs (part 1)
	13. Mini Quiz	13. Metals and acid (Demonstration)	13. Comparison of energy resources	13. Inertia and inertial mass ((higher only)	13. Sexual vs asexual reproduction	13. Developing new drugs (part 2)
9	14. Metals and non-metals	14. Metals and water (Demonstration)	14. Mini Quiz	14. Investigate Newton's Second Law of motion (R. Practical)	14. Examples of unusual reproduction	14. Scatter Graphs and Health
	15. Uses of metals	15. Redox reactions (Higher only)	15. Introduction to waves	15. Newton's third law	15. Inheritance (genetic cross diagrams)	15. Frequency tables and histograms
	16. Alloys	16. Acids and bases	16. Waves equation	16. Stopping distances	16. Family trees	16. Analysis data
	17. Properties and uses of alloys	17. Neutralisation	17. Measuring period of a wave	17. Energy transfers in stopping	17. Genetic diseases and sex determination	17. Mini Quiz
	18. Alkali metals (Demonstration)	18. RP: Soluble Salts	18. RP: Measuring speed of a wave using a ripple tank	18. Momentum (higher only)		
	19. Halogens	1. Reactivity series and displacement reactions (Practical)	19. Measuring the speed of sound	19. Hooke's Law		
	20. Noble Gases	2. Ionic half equations for displacement (Higher only)	20. EM Spectrum	20. Relationship between force and extension		
	21. Gas tests (Demonstration/Practical)	3. Reactivity series and extraction methods		21. Circular Motion		
		4. Electrolysis of molten compounds (ionic half equations - higher only)		22. Magnets		
		5. Electrolysis of aqueous compounds (ionic half equations - higher only)		23. Magnetic fields		
		6. Electrolysis part 1 (R.Practical)		24. Electromagnets		
		7. Electrolysis part 2 (R.Practical)				